

A Sensitive Method for the Rapid Determination of Underivatized Ethyl Carbamate in Fortified Wine by Liquid Chromatography-Electrospray Tandem Mass Spectrometry

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Abstract This work presents the optimization of a miniaturized liquid-liquid extraction (mLLE) followed by reversed-phase liquid chromatography-electrospray tandem mass spectrometry (RP-HPLC-MS/MS) for the determination of ethyl carbamate (EC) in fortified wine, without using derivatizing agents. The mLLE was optimized by an experimental design. Thus, 15 mL of wine and 8 mL of ethyl acetate were used for the extraction. After concentration, each extract was injected into the HPLC-MS/MS equipment and the characteristic secondary ion transition of EC ($m/z = 90.10 \rightarrow 62.05$) was used for the quantification purposes. The proposed method presented a good linearity ($R^2 = 0.9999$) and a high sensitivity with low limits of detection (LOD) and quantification (LOQ), 0.17 and 0.52 $\mu\text{g L}^{-1}$, respectively. The precision (repeatability and reproducibility) never exceeded 8% of variation, and the recoveries varied between 93 and 114%. The applicability of the method was checked through the analysis of 24 fortified wines, with EC values ranging between 23 ± 1 and $194 \pm 5 \mu\text{g L}^{-1}$. All chromatograms revealed good peak resolutions. This new method is efficient for the simple, fast, and reliable determination of EC in fortified wines, providing great sensitivity without using derivatizing agents or large volumes of organic solvents.

Keywords Ethyl carbamate · Liquid-liquid extraction · Wine · Liquid chromatography-mass spectrometry · Design of experiments

Abbreviations

EC	Ethyl carbamate
mLLE	Miniaturized liquid-liquid extraction
LC	Liquid chromatography
MS/MS	Tandem mass spectrometry
MRM	Multiple reaction-monitoring
FW	Fortified wine
DoE	Design of experiments
IS	Internal standard
ME	Matrix effect
LOD	Limit of detection
LOQ	Limit of quantification

Introduction

Ethyl carbamate (EC), also known as urethane (PubChem CID 5641), is the ethyl ester of carbamic acid that is formed in fermented foods and beverages, namely fortified wines, by the reaction of nitrogen compounds with ethanol (JECFA 2005). It is recognized as a toxic compound and was reclassified in 2007 by the International Agency of Research on Cancer (IARC) as probably carcinogenic to humans (group 2A) (Baan et al. 2007). The most relevant pathways for the formation of EC in alcoholic beverages have already been identified. The chemical family of EC's main precursors are cyanide and carbamoyl compounds, namely urea, carbamoyl phosphate, citrulline, and hydrogen cyanide (Jiao et al. 2014). In wines, EC is mainly formed during and after fermentation, by the reaction of carbamoyl compounds with ethanol. The

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most common way for EC to occur in acidic medium, such as wine, is the reaction of ethanol with urea. Urea and citrulline are both derived from the arginine metabolism during the fermentative processes (Arena et al. 1999; Stevens and Ough 1993; Weber and Sharypov 2009). External factors, such as temperature and pH influence the kinetics of these reactions (Jiao et al. 2014; Weber and Sharypov 2009). Its formation is also associated with storage time (Hasnip et al. 2004).

Canadian authorities, based on a study about EC toxicity, investigated by Schmahl et al. (1977), imposed for the first time, in 1985, legislation regulating the EC limit values in alcoholic beverages. Until now, a common legislation to regulate the maximum level of EC (EFSA 2007) has not been established in the European Union. However, some countries have established their own legislation. For example, in regard to fortified wines, the Czech Republic implemented an identical legislation to that of Canada, $100 \mu\text{g L}^{-1}$, while the USA are more restrictive with a limit of $60 \mu\text{g L}^{-1}$ (Weber and Sharypov 2009). The concerns raised by the toxicological aspects and the low concentration levels of EC found in alcoholic beverages have motivated several researchers to improve its analytical quantification (EFSA 2007; Weber and Sharypov 2009).

Gas chromatography coupled with mass spectrometry (GC-MS) has become the most used technique for the quantification of EC and propyl carbamate, while butyl carbamate (BC) and EC labeled with deuterium (^{13}C or ^{15}N isotopes) have been used as internal standards (IARC 2010; Weber and Sharypov 2009). Different approaches have been used to extract EC before the chromatographic separation. The official method of the International Organization of Vine and Wine involves a complex and time-consuming extraction procedure, using a diatomaceous earth solid-phase extraction column, prior to the GC-MS analysis (Bertrand and Hitos 1998). Thus, several efforts have been done to develop methodologies to determine EC without using long procedures and laborious analyses, combining precision and high sensitivity. In this regard, headspace solid-phase microextraction (HS-SPME) has been gaining great highlight (Lachenmeier et al. 2006; Perestrelo et al. 2010; Whiton and Zoecklein 2002). More recently, ultrasound-assisted emulsification-microextraction (USAEME) and microextraction by packed sorbent (MEPS) have also been reported as feasible and easy-to-use extraction techniques to determine EC in alcoholic beverages (Leça et al. 2014; Liao et al. 2013).

The LLE methods found in literature to determine EC use significant volumes of sample and organic solvents along with time-consuming extraction procedures or derivatization agents. Fauhl and Wittkowski (1992) determined EC by continuous LLE through a Soxhlet apparatus followed by GC-MS analyses. Most recently, EC derivatized with bis(trimethylsilyl)trifluoroacetamide was extracted by LLE and quantified by GC-MS (Xu et al. 2012). Valente et al. (2014) used salting-out assisted LLE to analyze

EC derivatized with 9-xanthidrol followed by a high-performance liquid chromatography with fluorescence detection. These have emerged as an attempt to improve the methods used for EC determination.

The aim of this study was the development of a simple, efficient, and sensitive method to quantify EC in fortified wines, without using derivatizing agents, minimizing the reagent consumption, and reducing the use of toxic organic solvents, through the optimization of a mLLE procedure combined with reversed-phase liquid chromatography-electrospray tandem mass spectrometry (RP-HPLC-MS/MS).

Materials and Methods

Chemicals and Samples

Ethyl carbamate (EC) was purchased from Acros Organics (Geel, Belgium), while butyl carbamate (BC), used as internal standard (IS), was obtained from Sigma-Aldrich (Steinheim, Germany). All standards had a purity grade higher than 97%. Absolute ethanol was purchased from Sigma-Aldrich, tartaric acid, formic acid, and methanol (UPLC grade) were from Panreac (Barcelona, Spain), and ethyl acetate was from Fisher Scientific (Leicestershire, UK). Ultra-pure water (type 1) was obtained from a Simplicity® UV apparatus from Millipore (Milford, MA, USA). Ethyl carbamate and BC 1 g L^{-1} stock solutions were rigorously prepared in ultra-pure water. Intermediate solutions of 50 mg L^{-1} in EC and 10 mg L^{-1} in BC were prepared in order to obtain the calibration solutions, which were then used to spike the synthetic wine (containing 6 g L^{-1} of tartaric acid, 18% of ethanol, and pH adjusted to 3.5, with a 1 M NaOH solution). Each calibration point was extracted in duplicate and injected twice, within the validation range $1.0\text{--}200 \mu\text{g L}^{-1}$. To test the method applicability, a sample set of 24 fortified wines, with different ages (up to 86 years old) and ethanol content between 18 to 20%, were analyzed.

Apparatus and Chromatographic Conditions

A Nexera X2 UHPLC system (Shimadzu, Kyoto, Japan) consisting of binary LC-30AD pumps, a DGU-20 A5 degassing unit, a CTO-20A column oven, and a SIL-30AC autosampler was used. The LC system was coupled with the Shimadzu triple-quadrupole mass spectrometer LCMS-8040, equipped with an ESI ionization module. Sample extracts were separated in the reversed-phase mode (RP) using a Kinetex C18 column, $150 \times 2.1 \text{ mm}$, $2.6 \mu\text{m}$, 100 \AA , from Phenomenex (Torrance, CA, USA) thermostated at 40°C . The injection volume was $5 \mu\text{L}$ and all samples were injected twice. A gradient elution with methanol (solution A) and ultra-pure water with 0.1% of formic acid (solution B) were used at a 0.4 mL min^{-1} flow rate. The gradient started with the solution A maintained at 5% for 4 min

and then increased to 30% in 2 min, then to 100% in 1 min. Finally, solution A was reduced to 5% in 3 min and held at 5% for 5 min to prepare the next injection. All eluents were filtered through a hydrophilic polypropylene 0.2 μm pore size membrane filter (Pall Corporation, Ann Arbor). The column effluent was directed to the detector between the 0.5 and 9.0 min, and the rest was discarded. The LCMS-8040 ESI was operated in the positive ion mode along with multiple reaction-monitoring (MRM) mode, acquiring data for a single secondary ion transition ($m/z = 90.10 \rightarrow 62.05$) for EC and ($m/z = 118.00 \rightarrow 62.05$) for BC, with an optimal collision energy of -10 and -9 eV, respectively. The desolvation line temperature was maintained at 250°C and the block heater at 400°C ; the nebulizing gas flow was 2.5 L min^{-1} and the drying gas flow was 17.5 L min^{-1} . The data acquisition and all peak integration processing were performed with the Labsolutions 5.7 software (Shimadzu).

mLLE Optimization

Ethyl acetate was chosen to develop the extraction procedure and was optimized by performing a full factorial design. Experimental design methodologies are increasingly more relevant in the optimization of quantification methodologies of various compounds, including EC (Leça et al. 2015, 2014; Machado et al. 2012; Zhang and Zhang 2008). To maximize the HPLC-MS/MS response of EC, two experimental variables, at three levels, were studied: the sample volume at 8, 10, and 15 mL and the extractant volume at 4, 5, and 8 mL. All experiments were made varying the sample and extractant volumes in random order according to Table 1. A commercial fortified wine was used for the method optimization. The data analysis was performed using the Matlab software, version R2016b, to estimate the factors—levels combination that ensure the maximization of the HPLC-MS/MS response of EC.

mLLE Optimized Procedure

In 50 mL PTFE centrifuge tubes, 8 mL of ethyl acetate (extractant solvent) was added to 15 mL of sample/standard solution spiked with 50 μL of internal standard (10 mg L^{-1} BC solution). After vigorous shaking in vortex for 5 min, the tube was centrifuged at 4400 rpm for 5 min to achieve the separation of the liquid phases. Finally, an aliquot of the upper phase was collected and evaporated using a small nitrogen flow. The residue was redissolved with mobile phase B to a final volume of 1 mL and filtered through Chromafil PTFE 0.2 μm syringe filters (Macherey-Nagel, Düren, Germany). Each sample/standard solution was extracted twice. Finally, 5 μL of extract was injected twice into the HPLC-MS/MS system.

Table 1 Experiments of the full factorial design

Number	Sample volume (mL)	Solvent volume (mL)
1	8	8
2	15	4
3	15	4
4	8	5
5	10	8
6	10	8
7	8	4
8	8	8
9	11.5	6
10	8	5
11	10	4
12	10	5
13	15	5
14	8	4
15	11.5	6
16	10	5
17	10	4
18	15	5
19	15	8
20	15	8

Method Validation

The mLLE/RP-HPLC-MS/MS method optimized for the determination of EC in fortified wines was validated in terms of selectivity, linearity, sensitivity, matrix effect, precision, and accuracy.

Selectivity was appraised by the analysis of five fortified wines to ensure the absence of chromatographic interferences at the retention time of EC ($m/z = 90.10 \rightarrow 62.05$) and BC ($m/z = 118.00 \rightarrow 62.05$) which could compromise its quantification.

The working standard solutions were prepared by spiking synthetic wine at eight different concentration levels of EC 1, 2.5, 5, 12.5, 25, 50, 100, and $200\text{ }\mu\text{g L}^{-1}$ with a 50 μL spike of BC (10 mg L^{-1}). The calibration curve was obtained by plotting the analyte peak area ratio (EC area/BC area) against the corresponding EC concentration.

The linearity and sensitivity were determined based on the linear regression results. Linearity was determined by the coefficient of correlation, R -squared (R^2), of the generated calibration curve. Sensitivity was evaluated by determining the limit of detection (LOD) and the limit of quantification (LOQ), as follows: $\text{LOD} = 3.3\sigma/b$ and $\text{LOQ} = 10\sigma/b$, with σ as the standard deviation of the y -intercept and b the curve slope (Şengül 2016; Vial and Jardy 1999).

The matrix effect (ME) was assessed through the percentage of the quotient between the slopes of the curves

obtained from the standards solutions in synthetic wine and the curves obtained by spiking a dry and a sweet fortified wines with the same concentration of EC, by the

following equation (Dong et al. 2015; Dong and Xiao 2017; Matuszewski et al. 2003; Taverniers et al. 2004; Xian et al. 2016):

$$\%ME = \left[\frac{(\text{slope of fortified wine calibration curve} - \text{slope of synthetic wine calibration curve})}{\text{slope of synthetic wine calibration curve}} \right] \times 100$$

Precision was estimated based on inter- and intra-day analyses of two standard solutions and one fortified wine. Intra-day repeatability was assessed by the response of ten successive extractions. Inter-day reproducibility was assessed by the results of the analyses of five extractions of the same samples in three different days. These parameters were expressed as relative standard deviation (%RSD).

The accuracy of the method was evaluated through a recovery study, by spiking a fortified wine in triplicate, with known amounts of EC at four representative concentration levels (25, 50, 100, and 200 $\mu\text{g L}^{-1}$), within the calibration range. Recovery was calculated by comparing the mean values of the three replicates with the theoretical concentrations of each one.

The method was applied to 24 different fortified wines, with ages between less than 1 and up to 86 years old, in order to confirm the applicability of the proposed RP-HPLC-MS/MS method for determining EC in fortified wines.

Results and Discussion

HPLC-MS/MS, combined with MRM, was selected for the quantification of EC in fortified wines due to its high loading capacity, sensitivity, and selectivity (Moein et al. 2016). The LC-MS/MS technique was previously used to quantify EC but typically with a derivatization step or other extraction techniques in sample preparation (Alberts et al. 2011; Deák et al. 2010). To optimize the analytical technique, a solution of EC and BC was directly injected into the HPLC-MS/MS equipment to determine their retention times, 1.9 and 7.8 min, respectively. The optimal collision energy to quantify EC with BC as internal standard was defined through multiple tests, using the Labsolutions 5.7 software. The elution gradient was optimized with real samples to obtain the best separation and peak resolution.

mLLE Method Optimization

From the previously survey for solvents free of EC, by Leça et al. (2014), acetonitrile and ethyl acetate were selected, due to their lower density, to firstly perform a QuEChERS (quick, easy, cheap, effective, rugged, and safe) extraction procedure. The QuEChERS approach was first reported by Anastassiades

et al. (2003) and is characterized as user-friendly, inexpensive, effective, robust, secure, and with high recoveries. This procedure was firstly chosen to extract the target compound from fortified wines since it has been widely applied to various classes of compounds in several matrices (Bruzzoniti et al. 2014). The influence of salts and buffers on the extraction yield was studied. Since they did not increase the extraction yield of EC, these were removed from the extraction procedure and the extraction was transformed into a mLLE, with the advantage of being cheaper and user-friendly. Acetonitrile and ethyl acetate were both tested. Ethyl acetate was chosen as the extractant solvent rather than acetonitrile, since it leads to chromatograms with less interferences and better peak shapes. Acetonitrile extracts increased background noise and originate interferences that coelute with EC peak.

To optimize the mLLE extraction, a full factorial design was carried out, considering two factors, the sample and extraction solvent volumes. The optimum conditions were those which maximize the EC chromatographic peak area in the HPLC-MS/MS equipment. The full factorial design was done considering three levels per factor, namely the sample volume at 8, 10, and 15 mL and the extractant volume at 4, 5, and 8 mL. The two variables analyzed revealed to be significant, as well as the interaction factor between them (p values lower than 0.05). As illustrated by Fig. 1, the maximum EC chromatographic peak area, for the mLLE method, can be achieved using 15 mL of fortified wine sample combined with 8 mL of ethyl acetate.

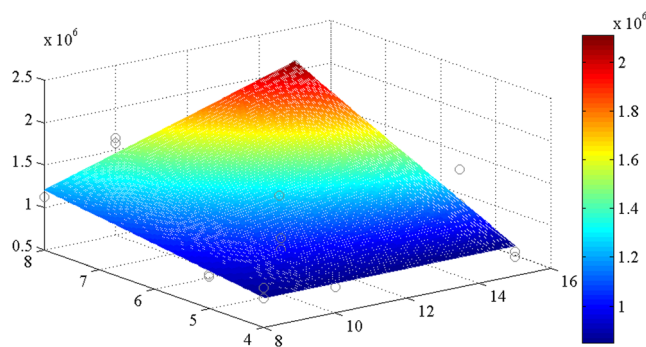


Fig. 1 Full factorial design to optimize the miniaturized LLE procedure with sample volume, extraction solvent volume, and the HPLC-MS/MS response as variables. The graphic illustrates the variation of the HPLC-MS/MS response

mLLE Method Validation

The proposed mLLE/RP-HPLC-MS/MS method to quantify EC in fortified wines showed good selectivity, verified by the absence of interferences at EC and BC retention times in synthetic and real wine samples, as depicted in Fig. 2.

It is recognized that the matrix effect can compromise the results of an analytical method, especially when the intention is to analyze samples of high complexity, such as fortified wines. Thus, the coefficient of variation between the slopes of the synthetic wine calibration curve and those of the spiked dry and sweet fortified wines was 3.4 and 2.5%, respectively (curves are almost parallel). These results show that the differences between the curves obtained using the synthetic and real wine matrices are negligible.

Table 2 summarizes the method validation results. A single calibration curve was generated by the average response of eight concentration levels, prepared in synthetic wine. Each level was represented by three extracts, injected twice. A good correlation coefficient was obtained, $R^2 = 0.9999$, confirming the linearity of the method. The LOD and LOQ results determined by the calibration curve, 0.17 and 0.52 $\mu\text{g L}^{-1}$, respectively, are close to and, in most cases, lower than those found in literature (Ajtony et al. 2013; IARC 2010; Jagerdeo et al. 2002; Lachenmeier et al. 2005; Lachenmeier et al. 2006; Leça et al. 2014; Mo et al. 2014; Perestrelo et al. 2010; Whiton and Zoecklein 2002; Xu et al. 2012; Zhang et al. 2014), conferring a good sensitivity to the

Table 2 Validation results obtained for the proposed mLLE/RP-HPLC-MS/MS method

Parameter	Result
Linear regression	$A_{\text{rel}} = 0.1706[EC] + 0.0438$
Linear concentration range	1.0–200 $\mu\text{g L}^{-1}$
R^2	0.9999
LOD ($\mu\text{g L}^{-1}$)	0.17
LOQ ($\mu\text{g L}^{-1}$)	0.52
Recovery	%
FW + 25 $\mu\text{g L}^{-1}$ of EC	93
FW + 50 $\mu\text{g L}^{-1}$ of EC	113
FW + 100 $\mu\text{g L}^{-1}$ of EC	113
FW + 200 $\mu\text{g L}^{-1}$ of EC	114

A_{rel} relative area (EC peak area/BC peak area), $[EC]$ EC concentration in $\mu\text{g L}^{-1}$, LOD limit of detection, LOQ limit of quantification, FW fortified wine, EC ethyl carbamate

developed method. The LOQ result is about ten times more sensible when compared to the direct determination approach ($\text{LOQ} = 5.0 \mu\text{g L}^{-1}$) used by Lijuan et al. (2012). The method also revealed a good repeatability (1.6–5.6%) and reproducibility (2.1–8.0%) and never exceeded 8% of RSD. In turn, the results of the recovery study ranged between 93 and 114%, as summarized in Table 2, demonstrating the accuracy of the method.

Regarding the applicability of the proposed mLLE/RP-HPLC-MS/MS method for the determination of EC in fortified

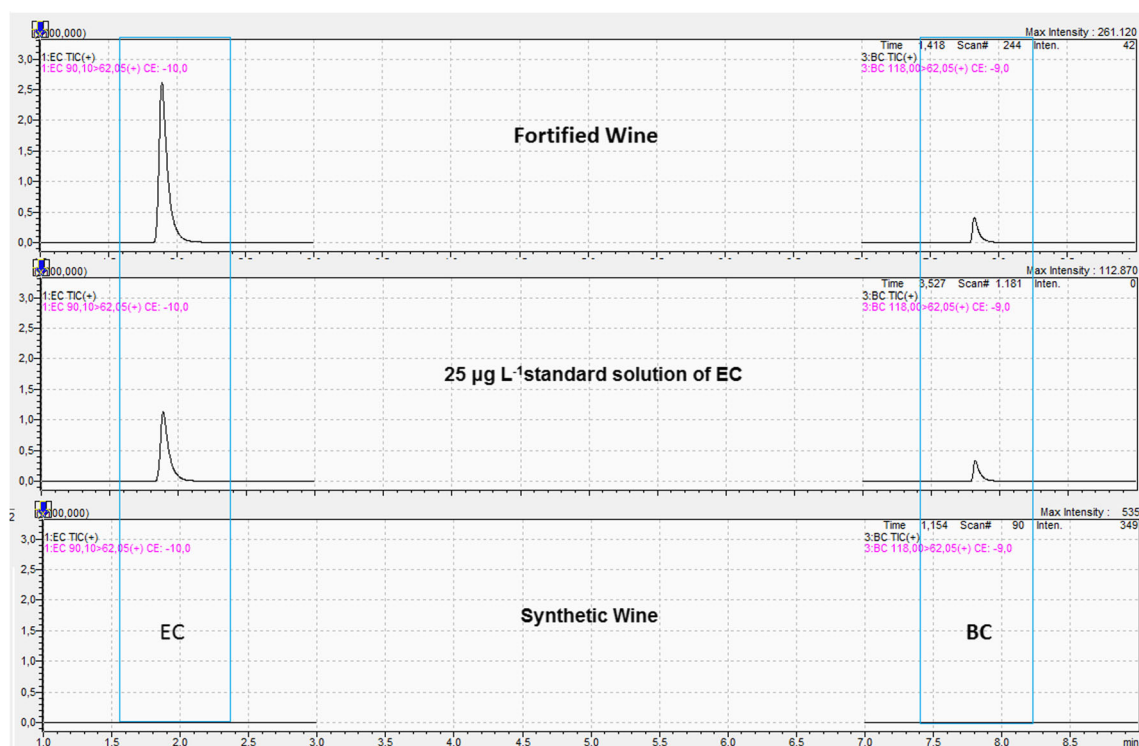


Fig. 2 Typical chromatograms of synthetic wine used to generate the calibration, 25 $\mu\text{g L}^{-1}$ standard solution and fortified wine. EC ethyl carbamate, BC butyl carbamate (internal standard)

Table 3 Application of the proposed mLLE/RP-HPLC-MS/MS method for the quantification of EC in 24 fortified wines of different ages

Fortified wine	Wine age	Concentration	SD
sample	(years)	($\mu\text{g L}^{-1}$)	($n = 6$)
FW1	–	194	5
FW2	–	42	2
FW3	–	26	1
FW4	<1	13.4	0.9
FW5	<1	16.4	0.3
FW6	<1	32.1	0.6
FW7	<1	49	2
FW8	3	24	1
FW9	3	23	1
FW10	5	89	2
FW11	3	43.9	0.5
FW12	3	57	2
FW13	3	110	4
FW14	3	103	3
FW15	3	98.7	0.9
FW16	3	23	1
FW17	3	24.7	0.2
FW18	5	131	7
FW19	5	77	2
FW20	5	84	2
FW21	5	75	2
FW22	17	150	3
FW23	20	61.8	0.9
FW24	86	177	11

FW fortified wine, SD standard deviation

wines, a set of 24 fortified wines were analyzed. The results are shown in Table 3. The resolution, selectivity, and precision of the obtained chromatograms and corresponding results confirm the applicability of the proposed method to quantify EC in fortified wines. The quantified concentrations varied from 23 ± 1 to $194 \pm 5 \mu\text{g L}^{-1}$, showing that the developed method covers the concentration range usually found for this compound in wines (Table 3). Despite the wide range of ages of the analyzed sample set, a direct correlation between EC values and wine age was not found. Six wines presented ethyl carbamate concentrations superior to $100 \mu\text{g L}^{-1}$ and 12 superior to $60 \mu\text{g L}^{-1}$.

Conclusion

A simple, sensitive, and eco-friendly method based on mLLE and RP-HPLC-MS/MS was successfully developed for the determination of ethyl carbamate in fortified wine samples, without resorting to derivatization. The mLLE optimization was achieved, ensuring a good compromise between sensitivity and low usage volumes. The proposed method shows good linearity,

sensitivity, selectivity, precision, and accuracy. The mLLE provides a favorable extraction of EC in less than 15 min and a LC-MS/MS analysis in 18 min. The method was successfully applied to determine the concentration of EC in 24 fortified wines confirming its suitability. This low-cost method is a great contribute for simplifying the EC determination in wines, for mitigation and control purposes. Future perspectives should also be focused on the simultaneous analyses of EC and its main precursors.

Compliance with Ethical Standards

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Conflict of Interest João M. Leça declares that he has no conflict of interest. Vanda Pereira declares that she has no conflict of interest. Ana C. Pereira declares that she has no conflict of interest. José C. Marques declares that he has no conflict of interest.

Ethical Approval This article does not contain any studies with human or animal subjects.

Informed Consent An informed consent is not applicable for the nature of this study.

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